

Fermenter Design for (Batch and Continuous Operation) Production of Ethanol from Corn

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Abstract: The liquid portion of the slurry has 8-12% ethanol by weight. The whole process requires 48-72 hours. The batch fermentation systems can be used and out of is more popular. For continuous operation we need 4 numbers of fermenter according to the fermentation time. The Batch and Continuous fermentation systems can be used and out of both batch process is more popular. By using material balance, we get the volume of fermenter for 1 KL ethanol production. With help of volume and design equations we can calculate Length, Height, Thickness, I.D., O.D., Weight of Vessel, Head thickness & cost of Fermenter. The sizes of bioreactor vary from microbial cell to shake flask (100-1000 ml) to laboratory scale vessel (1 – 50 L) to pilot level (0.3 – 10 m³) to industrial scale (2 -500 m³) for large volume industrial applications. Lab scale fermenter are typically constructed with the glass and the pilot scale and the industrial scale vessel are normally fabricated with stainless steel. 1000 ml of 0.1M sodium acetate buffer solution poured into 1000 ml (Erlenmeyer) conical flask, 20 g of *Aspergillus Niger* (crude enzyme), 5 g of brewer's yeast, 20 g of treated corn added and 1 g of MgSO₄, 2 g of (NH₄) H₂PO₄ added as nutrient. The flask corked properly, sealed with aluminium foil paper and incubated at 30 °C for 48-72 hrs. in an incubator. We can increase the fermentation time 4-6 days for analysis of yield of ethanol from biomass. As per literature study optimum time of fermentation is 48-72 hrs. i. e. 2-3 days. As per analysis 1000 ml (1 L) of ferment yield of 98.7 ml of ethanol. That shows the 9.87 % of yield of ethanol from corn with substrate concentration 140-160 gm/L. With help of yield we design fermenter (Reaction Vessel) for both continuous and batch operation. With help of design book and design equation we calculating the L, Di, t, h and Do of fermenter.

Keywords: Corn, Ethanol, Batch and continuous operation, Fermenter Design.

Introduction

Ethanol as most important alcohol can be produced by converting the sugar content of any starchy material into alcohol with the evolution of carbon dioxide (CO₂) under controlled environmental conditions. Production of ethanol from lignocellulose materials such as corncob, cornstalk, cornhusk, sugarcane bagasse and sugarcane bark though faces challenges but can substitute bio-ethanol production from edible food substances. In this process yeast and heat are used to break down complex sugars into more simple sugars, producing ethanol. Starchy materials are first hydrolysed to fermentable sugars and subsequently fermented with the required yeast species to produce ethanol. Ethanol is best alternative over non-renewable petroleum products while corn stalk is suitable agricultural waste for ethanol production. Corn stalks are initially pre-treated to make it more and productive for saccharification and fermentation to yield of ethanol.

1. Wet Milling

Wet milling is used to produce many products besides fuel ethanol. Large-scale, capital-intensive, corn processing wet mills produce such varied products as high fructose corn syrup (HFCS), biodegradable plastics, food additives such as citric acid and xanthan gum, corn oil (cooking oil) and livestock feed.

2. Dry-Grind

In dry mill process, the entire grain kernel is ground into flour. The starch in the flour is converted to ethanol during the fermentation process, creating carbon dioxide and distillers' grain. Fermentation is one of the oldest process known to man. Ethanol is mainly produced by dry-grinding process (approximately 67%) and its percentage is increasing rapidly. Dry-grind ethanol process whole grain is processed and the residual components are separated at the end of the process. There are five major steps in the dry-grind method of ethanol production.

Raw Materials and Chemicals

1. Biomass (Corn)
2. Enzymes
3. Yeast (Microorganism)
4. NaOH
5. Water
6. Nutrients for Microorganism

Unit Operation for Material Balance

1. Pre-treatment
2. Scarification and Fermentation
3. Filtration
4. Distillation and Recovery

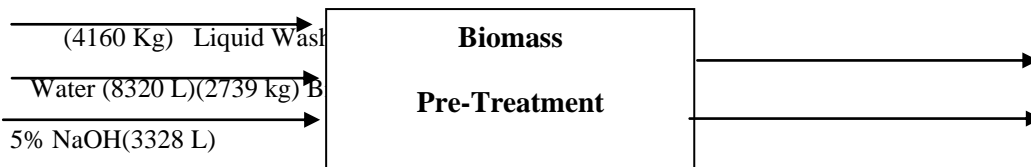
Material Balance

Material Balance for Pre-treatment

Basis – 1000 L/Day Ethanol Production.

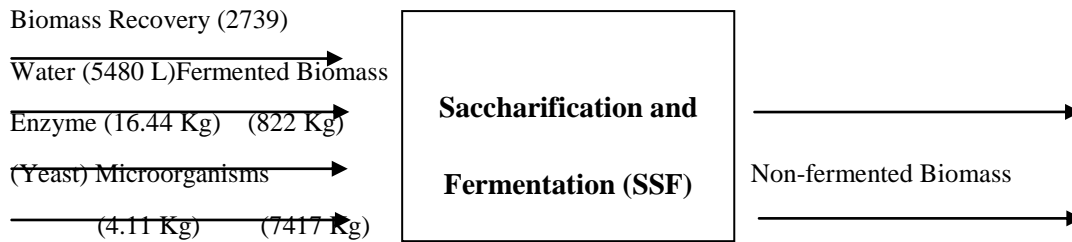
For production of ethanol from corn needs pre-treatment. Here, alkali treatment (5% of NaOH) are used for treatment of lignocellulose biomass for increase fermentation of biomass and increase yield of ethanol. Conditions for pre-treatment of biomass is 50 °C temperature and 2 hrs. Contact time. Biomass contains 53% cellulose, 19% hemicellulose, 15.5% lignin and 12.5% other materials. After biomass pre-treatment recoverable biomass feed to Simultaneous Saccharification and Fermentation (SSF) unit. According to composition 4160 kg of biomass contains 2204 kg cellulose, 790 kg hemicellulose, 649 kg lignin and 515 kg other components.

Biomass (Corn)



Material Balance for SSF

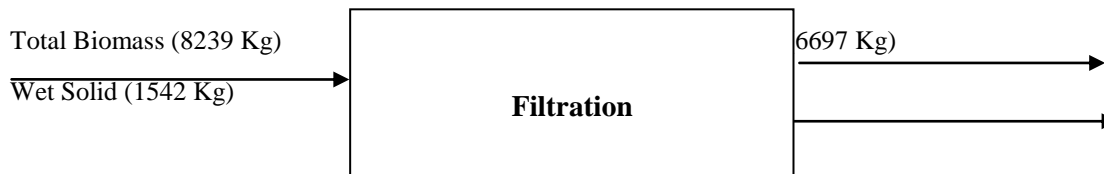
After pre-treatment of biomass is send in fermenter for SSF process where Saccharification and Fermentation. In scarification glucoamylase completes the breakdown of the starch into simple sugar. This process occurs at 30-35 °C. In the fermentation step yeast grown in seed tanks are added to the corn mash to begin the process of converting the simple sugars to ethanol. For fermentation requires 72 hrs. time and optimum temperature 35-40 °C.



According to above material balance, we get amount of ferment requires for production of 1000 L/Day ethanol. By using this volume we can calculate fermenter thickness, head thickness, inside and outside diameter, Length of fermenter by using design equations. The liquid portion of the slurry has 8-12% ethanol by weight. The whole process requires 48-72 hours. The Batch and Continuous fermentation systems can be used and out of both batch process is more popular.

Material Balance for Filter

Filtration process carried out for separation of filtrate (Liquid) from the solid- liquid mixture. Total biomass feed to the filter unit where liquid to be separated from the solid. Overall Material Balance for Filter,



$$\text{Total Biomass feed (8239 kg)} = \text{Filtrate (6697 kg)} + \text{Wet Solids (1542 kg)}$$

Material Balance for Filtrate,

$$\text{Filtrate (6697 kg)} = 822 \text{ kg Fermented Biomass} + 5480 \text{ kg Water} + 395 \text{ kg non-fermented biomass}$$

$$\text{Wet solid 1542 kg} = \text{Total Biomass Feed 8239 kg} - \text{Filtrate 6697 kg}$$

Material Balance for Distillation Column

After fermentation, the liquid portion of the slurry has 8-12% ethanol by weight. Because ethanol boils at a lower temperature than water does the ethanol can be separated by a process called distillation. Conventional distillation/rectification systems can produce ethanol at 92-95% purity. Maximum recovery of ethanol is 98 % of fermented biomass.

Overall Material Balance for Ethanol,

$$\text{Feed 6697 kg} = \text{Distillate 805} + \text{Bottom Product 5892 kg}$$

Here, fermented biomass in filtrate is 822 kg but maximum recovery of ethanol will be 98%. Hence, the distillate will be $0.98 * 822 = 805 \text{ kg}$

Distillate (805 Kg)



Volume of ethanol produce can be calculate by using density of ethanol (0.8 kg/L). As density is the ratio of mass / volume.

$$\text{Volume} = \text{mass} / \text{density}$$

$$\text{Volume of Ethanol (in L)} = 805 / 0.8 = 1006 \text{ L.}$$

$$\text{Rate of Bottom Product} = \text{Feed} - \text{Distillate} = 6697 - 805 = 5892 \text{ kg}$$

Fermenter Design

Introduction

Fermenter is a vessel that maintains optimum environment for the development of significant microorganism used in large scale fermentation process and the commercial production of products like Alcoholic beverages, Enzymes, Antibiotics, Organic acids etc. The fermenter aims to produce biological product like vaccines and hormones.

Laboratory Scale Bioreactor/Fermenter

In fermentation with strict aseptic requirements it is important to select materials that can withstand repeated sterilization cycles. For small scale operation glass and/or stainless steel. Glass is useful because it gives smooth surfaces is non-toxic, corrosion proof and it is usually easy to examine the interior of vessel. The glass should be 100% borosilicate. The laboratory bioreactor can be made from,

1. Glass bioreactor (without the jacket) with an upper stainless steel lid.
2. Glass bioreactor (with the jacket) with an upper stainless steel lid.
3. Glass bioreactor (without the jacket) with the upper and lower stainless steel lids.
4. Two part bioreactor - glass/stainless steel
5. Stainless steel bioreactor with peepholes.

Fermenter Design

Stepwise Procedure for Fermenter Design

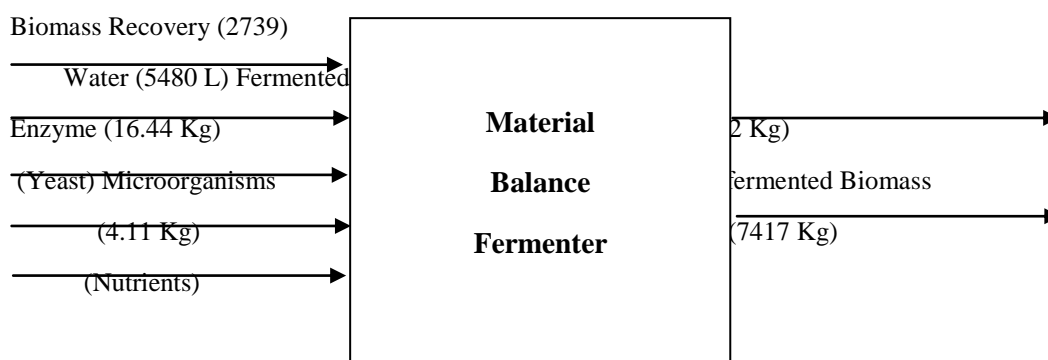
1. Calculate the volume of ethanol produce from 1 or 2 L of ferment by lab based experimental analysis.
2. Ferment means the medium and molasses mixture which produce the citric acid.
3. Convert that volume for 1 T/Day citric acid production i.e. by calculation we get volume of ferment requires to produce 1 T/day ethanol.
4. By considering volume we can calculate internal diameter and length of fermenter.
5. Then using design equation we can calculate outer diameter & thickness of vessel.
6. With help of outer diameter and thickness next we can calculate the weight of heads and vessel.
7. Finally we can calculate the cost of vessel requires. (Total cost for vessel = Material cost and Fabrication Cost)
8. Then add cost of agitator, motor, supports, gas Spurger etc.

Feed Conditions

Temperature – $35 \sim 40^{\circ}\text{C}$ and Pressure – 101.325 KPa (1 atm)

Fermentation Time – 48-72 hrs.

Fermentation Balance



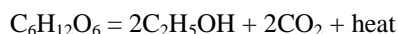
Here, fermented biomass in filtrate is 822 kg but maximum recovery of ethanol will be 98%. Hence, the distillate will be $0.98 * 822 = 805$ kg

Here, fermented biomass in filtrate is 99 kg but maximum recovery of ethanol will be 98%.

Hence, the distillate will be $0.98 * 99 = 98$ gm. Volume of ethanol produce can be calculate by using density of ethanol (0.875 gm/ml). As density is the ratio of mass / volume. Volume of Ethanol (ml) = $97/0.875 = 110$ ml/L of Ferment.

Fermentation Kinetics

In the anaerobic pathway, glucose is converted to ethanol and carbon dioxide via glycolysis. The overall reaction produces two moles of ethanol and carbon dioxide for every mole of glucose consumed, with the reaction energy as two moles of ATP for use in biosynthesis or maintenance.



As per various literature the conversion of glucose or fructose to ethanol will be up to 75 % and our ferment contains the 19.15 % of fructose that can be converted in to the ethanol. Total ferment feed to fermenter will be 8239 kg/day. Ferment contains distilled water, corn, nutrients, yeast etc.

According to material balance contains the 19.15 % of fructose that will be 1577.75 kg/day of fructose. Mass flow rate of glucose will be 65.75 kg/hr.

Molar Flow rate of Fructose = Mass Flow Rate/ Molecular Weight of Fructose

Molecular Wt. Fructose ($C_6H_{12}O_6$) = $12*6 + 12*1 + 16*6 = 180$ kg /Kmol

(F_{Ao}) Molar Flow Rate for Glucose = $65.75/180 = 0.365$ Kmol/hrs.

Density of Fructose = 1694 kg/m^3

Density = Mass/Volume

Hence, Volume = Mass / Density

Volume = $1577.75 \text{ kg}/1694 \text{ kg/m}^3 = 0.9313 \text{ m}^3 = 931.2 \text{ L}$ Fructose in the ferment feed.

Volumetric Flow = Discharge / Time

$$= 931.2 \text{ L} / 24*3600 \text{ Sec}$$

Volumetric Flow = 0.01077 L/sec

Moles of Fructose in feed(N_{Ao}) = Mass of fructose/ Molecular Weight

$$= 1577.75/180$$

Total Moles of Fructose feed (N_{Ao}) = 8.65 kmol

Final Moles (N_A) = (N_{Ao}) (1- X_A)

Final Moles (N_A) = 2.16 kmol .

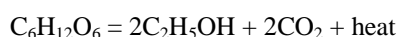
Initial Concentration of Fructose (C_{Ao}) = $n/V = 8.65*1000/931.2 \text{ L}$

$C_{Ao} = 9.28 \text{ mole/l or kmol/m}^3$

$C_A = 9.28 (1-X_A) = 9.28 (1-0.75) = 2.32 \text{ mole/lit}$

Where, C_{Ao} and C_A be the initial and final concentrations for fructose.

For first order kinetics,



Rate of Reaction – (r_A) = $K C_A = dC_A/dt = (C_{Ao} - C_A) / t$

$$-(r_A) = K C_{C_6H_{12}O_6} = dC_A/dt = (C_{A0} - C_A) / t$$

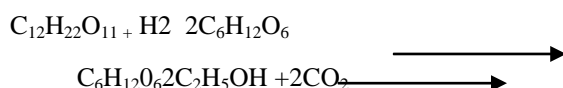
$$K * 2.32 \text{ (mole/l)} = 9.28 - 2.32 \text{ (mole/l)} / 24 \text{ hrs.}$$

$$\text{Hence, } K \text{ (Rate Constant)} = 0.1242 \text{ hr}^{-1}$$

Enthalpy and Gibbs Energy Data

SR.NO	COMPOUNDS	H (KJ/kmol)	G (KJ/kmol)
1	C ₆ H ₁₂ O ₆	-1274.45	-910.52
2	C ₂ H ₅ OH	-276.98	-174.138
3	CO ₂	-393.51	-394.38
4	C ₁₂ H ₂₂ O ₁₁	-2222.12	-1544.65
5	H ₂ O	-285.840	-228.61

Reactions involved in the manufacture of ethanol are as,

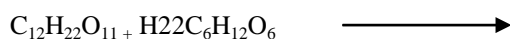


Total Moles of Fructose feed (N_{A0}) = 8.65 kmol

Moles of Fructose Reacted = 8.65 * Conversion = 8.65 * 0.75 = 6.487 kmol.

Moles of Fructose Reacted = 6.487 kmol.

For First Reaction,



As per stoichiometry requirement,

1 Moles Reacted of C₁₂H₂₂O₁₁ = 2kmol of C₆H₁₂O₆ Produce

X Moles Reacted of C₁₂H₂₂O₁₁ = 6.487 moles of C₆H₁₂O₆ Produce

Moles of C₁₂H₂₂O₁₁ required = 3.24 kmol

Moles of H₂ required = 3.24 kmol

For Reaction 1,

$$\Delta H^\circ_R = \Delta H^\circ_{\text{PRODUCTS}} - \Delta H^\circ_{\text{REACTANTS}}$$

$$= (6.48 \times (-1274.45)) - ((-2222.12 \times 3.24) + (-285.84 \times 3.24))$$

$$\Delta H^\circ_R = -135.36 \text{ KJ/kmol}$$

Heat of reaction is negative so reaction is exothermic.

$$\Delta G^\circ = \Delta G^\circ_{\text{PRODUCTS}} - \Delta G^\circ_{\text{REACTANTS}}$$

$$= (6.48 \times (-910.65)) - ((-1544.50 \times 3.24) + (-228.54 \times 3.24))$$

$$\Delta G^\circ = -156.36 \text{ KJ/kmol}$$

For Reaction – 2,

As per Reaction,

1 kmol Reacted of $C_6H_{12}O_6 = 2$ kmol of C_2H_5OH Produce

6.48 kmol Reacted of $C_6H_{12}O_6 = X$ kmol of $C_6H_{12}O_6$ Produce

Kmol of $C_6H_{12}O_6$ Produce = 12.96 kmol.

Similarly,

Kmol of CO_2 Produce = 12.96 kmol.

$$\begin{aligned}\Delta H^\circ_R &= \Delta H^\circ_{\text{PRODUCTS}} - \Delta H^\circ_{\text{REACTANTS}} \\ &= (12.96 \times (-276.98) + 12.96 \times (-393.51)) - 6.48 \times (-1274.45)\end{aligned}$$

$$\Delta H^\circ_R = -431.11 \text{ KJ/kmol}$$

Heat of reaction is negative so reaction is exothermic.

$$\begin{aligned}\Delta G^\circ &= \Delta G^\circ_{\text{PRODUCTS}} - \Delta G^\circ_{\text{REACTANTS}} \\ &= (12.96 \times (-174.38) + 12.96 \times (-394.38)) - (-910.52 \times 4.48)\end{aligned}$$

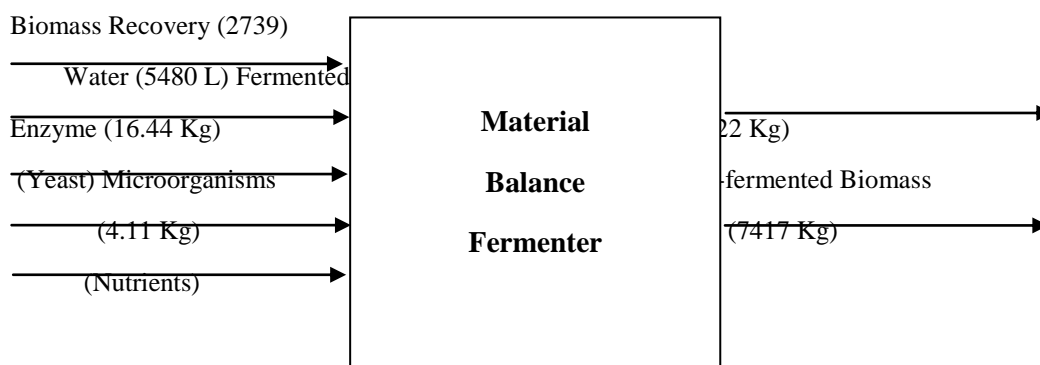
$$\Delta G^\circ = -1470.96 \text{ KJ/kmol}$$

Fermenter Design for Batch and Continuous Operation**Design of Fermenter for Batch Operation****Feed Conditions**

Temperature – 35 ~ 40 °C (For both Temp.)

Pressure – 101.325 KPa (1 atm)

Fermentation Time – 48 hrs.

Fermentation Balance

Here, fermented biomass in filtrate is 822 kg but maximum recovery of ethanol will be 98%. Hence, the distillate will be $0.98 \times 822 = 805$ kg

Here, fermented biomass in filtrate is 99 kg but maximum recovery of ethanol will be 98%. Hence, the distillate will be $0.98 \times 99 = 98$ gm. Volume of ethanol produce can be calculate by using density of ethanol (0.875 gm/ml). As density is the ratio of mass / volume. Volume of Ethanol (ml) = $97/0.875 = 110$ ml/L of Ferment.

Volume of Fermenter

As per material balance,

Total Feed (Composition) for Fermenter or Reactor = 8239.65 kg/Day

Total mass of Ethanol produce = 805.87 kg

8239.65 kg Feed ferment = 805.87 kg Ethanol output

Ethanol contains in slurry or ferment = $(805.87/8239.65) * 100 = 9.78 \%$

Design based on 1 Tonnes per day Production.

Density of Ethanol = 785 kg/m^3 at room temperature.

Using, Density = Mass/ Volume

Volume of Ethanol for 1T (1000 kg) production = $1000/785 = 1.27 \text{ m}^3$

= $1.27 * 1000 = 1270 \text{ L}$

Volume of Ethanol per produce batch = 1.27 m^3 (1270 L)

Volume of ferment can be calculated,

100L Ferment = 9.78 lit Ethanol (Fermentation Medium Produce 9.78 % of Ethanol)

X litre ferment = 1270 litre

X = 12985.86 L

For production of 1 T/Day requires 12985.86 L (12.99 m^3) ferment medium of corn.

For batch operation we need to 48 hrs. Time fermentation.

So, Volume will be twice = $1270 * 2 = 2570 \text{ L}$ of Ethanol.

For production of 2 T requires $12986 * 2 \text{ L} = 25976 \text{ L}$ of ferment medium.

Volume of Fermenter (Reactor) (20% Extra) = 31171 liters (31.18 m^3)
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Diameter and Length of Reactor (Fermenter)

Fermenter/Reactor cylindrical vessel can be design from process & mechanical design, Volume = 31.78 m^3

$$V = \text{Area} * \text{Length} = \pi/4 * D_i^2 * L$$

$$L = 1.5 D_i$$

$$V = \pi/4 * D_i^2 * 1.5 * D_i$$

$$31 = \pi/4 * D_i^3 * 1.5$$

By solving we get,

$$D_i = 5 \text{ m (take 5 m)}$$

$$L = 5 * 1.5 = 7.5 \text{ m}$$

$D_i = 5 \text{ m}$ and $L = 7 \text{ m}$
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Thickness and Outer Diameter of Fermenter

Shell Thickness (t_s)

$$t_s = P D_i / (2 f j - P) + C$$

$$= 101325 * 5 / (2 * 138 * 10^6 * 0.85 - 101325) + 1 \text{ mm}$$

$$= 0.001727 \text{ m} + 1 \text{ mm}$$

$$= 0.002160 * 1000 + 1 \text{ mm} = 2.160 \text{ mm} + 1 \text{ mm} = 3.160$$

$$t_s = 4 \text{ mm (Take)}$$

Outer Diameter (Do)

$$Do = Di + t_s = 5 + 2 * 4 * 10^{-3} = 5.08 \text{ m}$$

Here,

$$f = \text{Permissible stress for SS-316} = 138 \text{ MPa} = 138 * 10^6 \text{ Pa (20000 psi)}$$

$$J = \text{Joint Efficiency} = 0.85 \text{ (85\%)}$$

C = Corrosion Allowance

$$t_s = 4 \text{ mm and Do} = 5.06 \text{ m}$$

Weight of Cylindrical Shell and Tori Spherical Head

$$\text{Weight of Cylindrical Shell} = \pi/4 * (Do^2 - Di^2) * L * \rho \text{ (Density)}$$

$$\rho = 7950 \text{ kg/m}^3 \text{ (Density of Stainless Steel-316)}$$

$$\text{Weight of cylindrical shell} = \pi/4 * (5.06^2 - 5^2) * 7.5 * 7950$$

$$= 28266 \text{ kg}$$

$$\text{Weight of Tori Spherical Head} = \pi/4 * (Do^2 - Di^2) * \rho$$

$$= \pi/4 * (4.06^2 - 4^2) * 7950$$

$$\text{Weight of tori spherical head} = 3769 \text{ kg}$$

$$\text{Weight of Top and Bottom head} = 2 * 3019.5 = 7538 \text{ kg}$$

Total Weight of Vessel = Weight of Vessel + Weight of Head

$$= 28266 + 7538 = 35804 \text{ kg}$$

$$\text{Total Weight of Vessel} = 35804 \text{ kg}$$

Fermenter Costing

As per design total weight of material = 35804 kg.

Material cost = 250 Rs / kg

Total Material cost = 35804 * 250 = 89, 51,000 Rs.

Fabrication Cost (45-60 Rs / kg) = 60 * 35804 = 2848240 Rs

Agitator with motor and assembly cost = 2 Lac (Apr.)

Spurger cost = 0.5 Lac

Support cost is 15% of total material cost = 0.15 * 89, 51,000 Rs.

$$= 1342650 \text{ Rs}$$

Jacket cost is 15% of Total Material Cost = 1342650 Rs

$$\text{Total Fermenter Cost of fermenter} = \text{Rs. 1, 40, 34,540}$$

For complete fermentation of corn required time is 48 hrs. i.e. is 2 days so, in case of batch operation we need to install 2 No. of fermenter. 2 fermenter for 4 days operation including one day cleaning and feeding. According to fermentation time we feed material in 1st and 2nd feed 1st and 2nd day will get output as 3rd, and 4th with twice volume as continuous operation.

Total Fermenter Cost of fermenter = 1, 40, 34,540* 2 = 2, 80, 89,080 Rs.

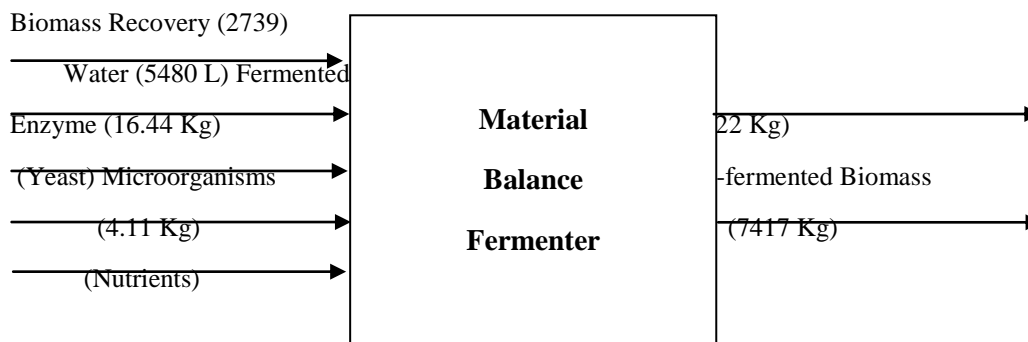
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Temperature – 35 ° 40 °C

Pressure – 101.325 KPa (1 atm)

Fermentation Time – 48-72 hrs.

Fermentation Balance



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Here, fermented biomass in filtrate is 99 kg but maximum recovery of ethanol will be 98%. Hence, the distillate will be $0.98 * 99 = 98$ gm. Volume of ethanol produce can be calculate by using density of ethanol (0.875 gm/ml). As density is the ratio of mass / volume. Volume of Ethanol (ml) = $97/0.875 = 110$ ml/L of Ferment.

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Volume of Ethanol per produce batch = 1.27 m³ (1270 L)

Volume of ferment can be calculated,

100L Ferment = 9.78 lit Ethanol (Fermentation Medium Produce 9.78 % of Ethanol)

X litre ferment = 1270 litre

X = 12985.86 L

For production of 1 T/Day requires 12985.86 L (12.99 m³) ferment medium of corn.

Volume of Fermenter (Reactor) (20% Extra) = 15580 liters (15.58 m³)

Diameter and Length of Reactor (Fermenter)

Fermenter cylindrical vessel can be design from process & mechanical design, Volume = 15.58 m³

$$V = \text{Area} * \text{Length} = \pi/4 * D_i^2 * L$$

$$L = 1.5 D_i$$

$$V = \pi/4 * D_i^2 * 1.5 * D_i$$

$$15.58 = \pi/4 * D_i^3 * 1.5$$

By solving we get,

$$D_i = 3.38 \text{ m} \approx 4 \text{ m}$$

$$L = 4 * 1.5 = 6 \text{ m}$$

$D_i = 4 \text{ m}$ and $L = 6 \text{ m}$

Thickness and Outer Diameter of Fermenter

Shell Thickness (t_s)

$$t_s = P D_i / (2 f j - P) + C$$

$$= 101325 * 4 / (2 * 138 * 10^6 * 0.85 - 101325) + 1 \text{ mm}$$

$$= 0.001727 \text{ m} + 1 \text{ mm}$$

$$= 0.001727 * 1000 + 1 \text{ mm} = 1.727 \text{ mm} + 1 \text{ mm}$$

$$\mathbf{t_s = 3 \text{ mm}}$$

Outer Diameter (D_o)

$$D_o = D_i + t_s = 4 + 2 * 3 * 10^{-3} = 4.06 \text{ m}$$

Here,

f = Permissible stress for SS-316 = 138 MPa = $138 * 10^6 \text{ Pa}$ (20000 psi)

J = Joint Efficiency = 0.85 (85%)

C = Corrosion Allowance

Weight of **$t_s = 3 \text{ mm}$ and $D_o = 4.06 \text{ m}$** head

$$\text{Weight of Cylindrical Shell} = \pi/4 * (D_o^2 - D_i^2) * L * \rho \text{ (Density)}$$

$$\rho = 7950 \text{ kg/m}^3 \text{ (Density of Stainless Steel-316)}$$

$$\text{Weight of cylindrical shell} = \pi/4 * (4.06^2 - 4^2) * 6 * 7950$$

$$= 18117 \text{ kg}$$

$$\begin{aligned}\text{Weight of Tori Spherical Head} &= \pi/4 * (D_o^2 - D_i^2) * \rho \\ &= \pi/4 * (4.06^2 - 4^2) * 7950\end{aligned}$$

Weight of tori spherical head = 3019 kg

$$\text{Weight of Top and Bottom head} = 2 * 3019.5 = 6039 \text{ kg}$$

Total Weight of Vessel = Weight of Vessel + Weight of Head

$$= 18117 + 6039 = 24156 \text{ kg}$$

Total Weight of Vessel = 24156 kg

Fermenter Costing

As per design total weight of material = 24156 kg.

Material cost = 250 Rs / kg

Total Material cost = 24156 * 250 = 60,39,000 Rs

Fabrication Cost (45-60 Rs / kg) = 60 * 24156 = 1449360 Rs

Agitator with motor and assembly cost = 2 Lac (Appr.)

Spurger cost = 0.5 Lac

Support cost is 15% of total material cost = 0.15 * 6039000 Rs. = 9,05,850 Rs

Jacket cost is 15% of Total Material Cost = 9,05,850 Rs

Total Fermenter Cost with Design Structure = 9550060 Rs

For complete fermentation of corn required time is 48-72 hrs. i.e. is 3 days So, for this operation in case of continuous operation we need to install 4 No. of fermenter. 3 fermenter for 3 days operation and one as standby. According to fermentation time we feed material in 1st, 2nd and 3rd as randomly 1st, 2nd and 3rd day will get output as 4th, 5th and 6th day. After it will continuous in manner. Actually process is batch but operation can operate at continuous in manner.

Hence as per operation requirement,

Hence,

Total cost of fermenter 4 no. of fermenter equal to 4* 9550060 = 3,82,00,240 Rs

Conclusion

The Batch and Continuous fermentation systems can be used and out of both batch process is more popular. By using material balance, we get the volume of fermenter for 1 KL ethanol production. With help of volume and design equations we can calculate Length, Height, Thickness, I.D., O.D., Weight of Vessel, Head thickness & cost of Fermenter. The sizes of bioreactor vary from microbial cell to shake flask (100-1000 ml) to laboratory scale vessel (1 – 50 L) to pilot level (0.3 – 10 m³) to industrial scale (2 -500 m³) for large volume industrial applications. Lab scale fermenter are typically constructed with the glass and the pilot scale and the industrial scale vessel are normally fabricated with stainless steel.

For continuous operation, for complete fermentation of corn required time is 48-72 hrs. I.e. is 3 days so, in case of continuous operation we need to install 4 No. of fermenter. 3 fermenter for 3 days operation and one as standby. According to fermentation time we feed material in 1st, 2nd and 3rd as randomly 1st, 2nd and 3rd day will get output as 4th, 5th and 6th day. After it will continuous in manner. Actually process is batch but

operation can operate at continuous in manner. For this system we design fermenter for 13.85 m^3 volume get internal and outer diameter as 4 and 4.06 m resp., Thickness and length 3mm and 6 m resp. We need four fermenter of volume 15.58 cubic meter each and need of total cost for both fermenter will be Rs. 3, 82, 00240.

For batch operation for complete fermentation of corn required time is 48-72 hrs. i.e. is 2-3 days so, in case of batch operation we need to install 2 No. of fermenter. 2 fermenter for 4 days operation including one day cleaning and feeding. According to fermentation time we feed material in 1st and 2nd feed 1st and 2nd day will get output as 3rd, and 4th with twice volume as continuous operation. For this system we design fermenter for 31.17 m^3 volume get internal and outer diameter as 5 and 5.06 m resp., Thickness and length 4 mm and 7.5 m resp. We need two fermenter of volume 31.17 each cubic meter each and need of total cost for both fermenter will be Rs 2, 80, 89,080. According to the cost view the batch operation is economical and requires low maintenance, labour cost and less time consuming and needs two fermenter. For continuous operation need high cost than the batch need four fermenter. In view of continuous is suitable operation than the batch in which regular feeding of fermenter and regular output. Also one fermenter is in standby operation

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